Macintosh HD:Users:simonem:MyResearch:DEPRO:HF(H2O)7:Fig-pKa.pdf

Figure 2: pKa *vs* T (more negative pKa means that the acids gets stronger). For reference, we also report experimental data for HF dissociation in bulk water (see Ayotte *et al*, J. Phys. Chem., 123, 184501 (2005) and references cited therein)